### **Oh Soy Good!**

# Ice Cream in a Bag

## Standard Laboratory Operating Procedure #1101

Laboratory: Biotechnology

Location: Food Science Lab

SOP prepared by: R. Sanders

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Is ice cream a complete food? Does it contain macromolecules that our cells need? Is soymilk a good alternative for people with lactose allergy? What are the most common types of milk available globally? Which type of milk is a good alternative for cultures that do not have proper refrigeration?

Safety: Safety glasses if working in a Food Science Lab

### Materials

1/2 cup (120 ml) milk variety

4 cups (960 ml) Ice

1/2 teaspoon (2.5 ml) Vanilla

4 tablespoons (60 g) Salt

1 tablespoon (15 g) Sugar

Quart (1L) Zip-loc bag

Gallon (4L) Zip-loc bag

Temperature probe

Scale

#### Procedure

- 1. Pour 120mL of milk into a quart size bag, using temperature probe measure the initial temperature of the milk sample and record in data table.
- 2. Add 2.5 mL of vanilla, and 15g of sugar in the quart size bag, with the milk sample.
- 3. Seal bag, squeezing out as much air as possible.
- 4. Pour 960mL of ice into the gallon size bag and then add 60g of salt on top of the ice.
- 5. Place the sealed quart size bag containing the milk solution in the gallon size bag containing the ice and salt mixture.
- 6. Seal, squeezing out as much air as possible.
- 7. Shake the gallon bag until milk solution in the quart size bag is frozen, about 5-10 minutes.
- 8. Use a temperature probe to measure the final temperature of the frozen milk sample and record in data table.
- 9. Use the following equation to calculate the freezing point depression:

Final Temperature – Initial Temperature =  $\Delta$  Tf (change in freezing point)

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